

Thermo Fisher S C I E N T I F I C

Introduction to FTIR and Raman Spectroscopy
Theory and Practice

2017 Spectroscopic Solutions Seminar

The Power of Molecular Spectroscopy

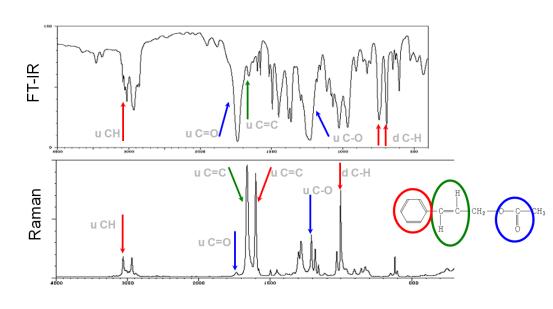


- Little or no sample prep, rapid screening
- ...on a range of samples...
 - Organics, inorganics, films, coatings, paper, fibers, more
- ...and answer basic questions...
 - What is this stuff? How much is in there? In what form?
 - ...or advanced questions.
 - What is the film thickness? What is the orientation?
 How do viscoelasticity and chemistry relate?

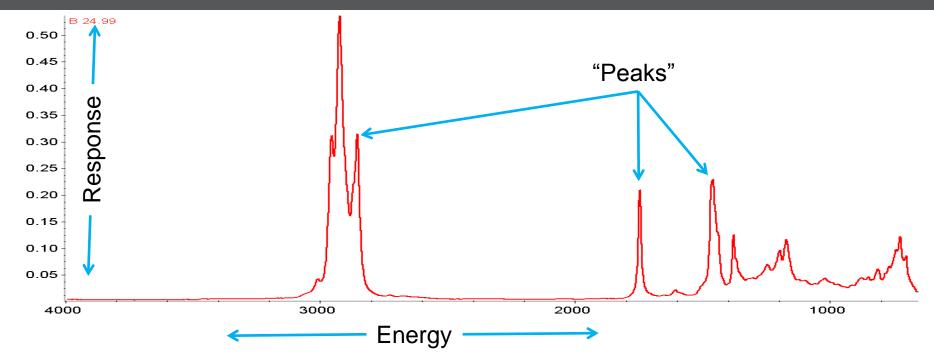


Vibrational Spectroscopy: FTIR and Raman

- Infrared Spectroscopy
 - Energy is absorbed directly...
 - ...which results in peaks
- Raman Spectroscopy
 - High energy light strikes sample...
 - ...the sample 'takes' some of the energy...
 - · ...which results in peaks
- Different methods but similar insights
 - The peaks provide information about the composition, concentration and more
 - Both are reporting upon vibrational motion of the molecules



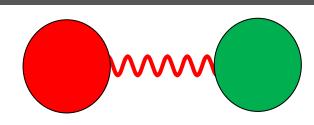
Basics: Information in the Spectrum



- Energy varies along the x-axis
 - Expressed in wavelength (nanometers, microns) or Wavenumber (cm⁻¹)
- Response (Intensity) is along the y-axis
 - Absorbance, transmittance, specular or diffuse reflectance, Raman scattering
- Peaks occur because different molecules interact with light differently
 - This is what enables spectroscopy to tell us something!



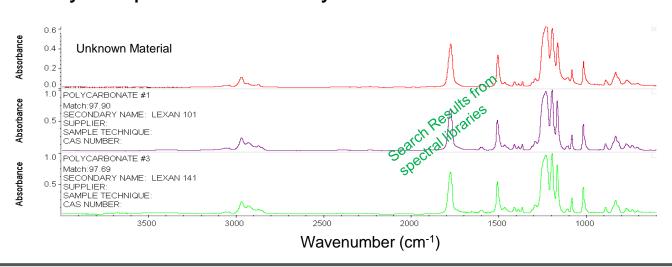
Basics: Spectroscopy and Composition



- Chemical bonds vibrate like springs
- Absorption occurs at the specific frequencies of those vibrations

Frequency of vibration =
$$\frac{1}{2\pi} \sqrt{\frac{k}{\mu}}$$
 Relates to bond strength Relates to mass

- As bond strength increases, frequency increases (C≡C > C=C > C-C)
- As mass decreases, frequency increases (C-H > C-C > C-Cl > C-Fe)
- Which is why the peaks tell a story about the material!



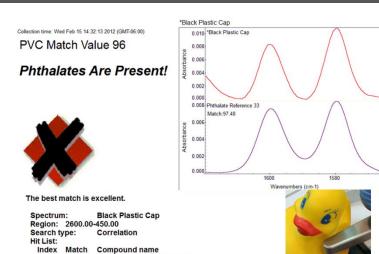


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Fourier Transform Infrared Spectroscopy

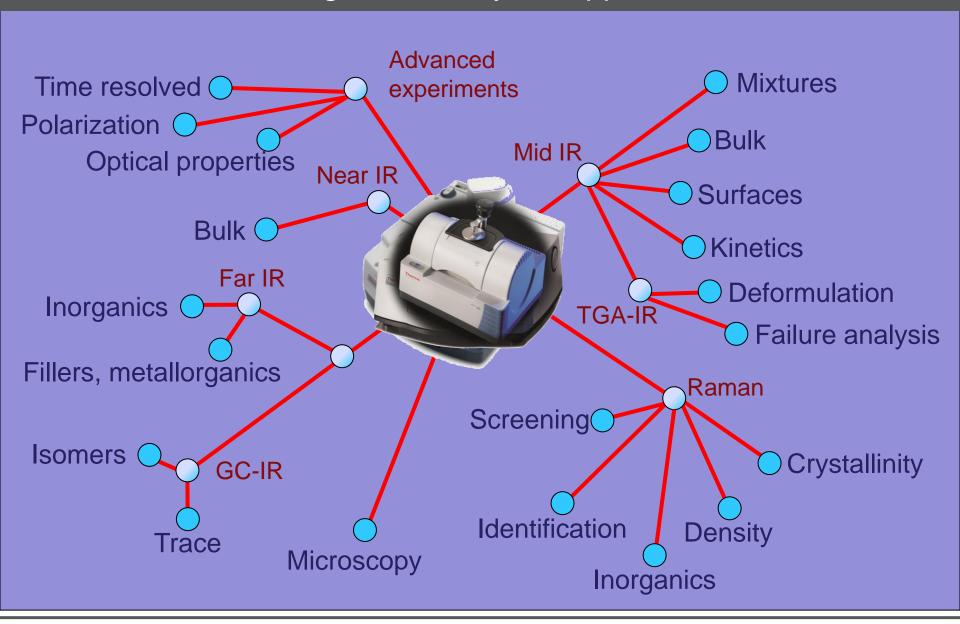
Basics: FTIR as a Laboratory Tool

- FTIR is used to solve problems
 - What is it? How much is in there?
- The ultimate Triage Technique
 - Get in...get an answer...act on it fast
- You generally need to accessorize
 - Add sampling tools, like ATR, Transmission
 - We will talk to that in the next presentation
- What do you need?
 - Single action, repeated analyses?
 - Flexibility?
 - Microscopy?

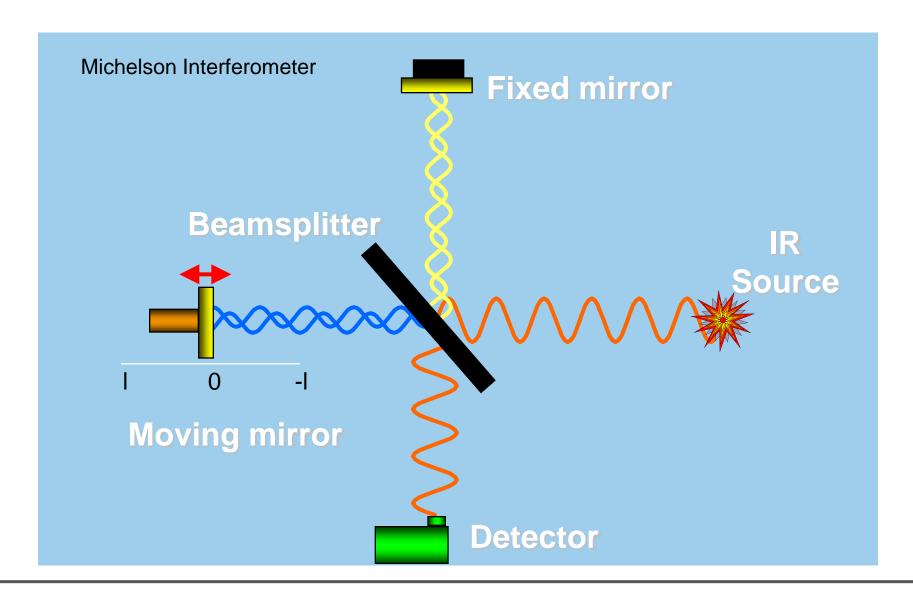




Basics: Select the Right Tool for your Application



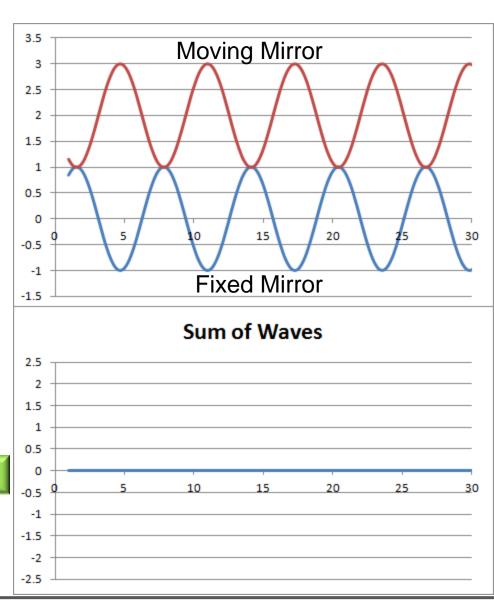
Basics: How FT-IR Works



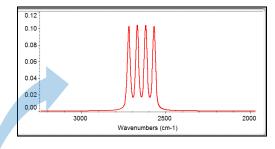
Time Domain to Frequency Domain

- One mirror moves...
- ...one doesn't
 - This causes the two beams to 'interfere' when they recombine
 - Watch how the detector signal (bottom) varies with mirror movement (top)...
 - The result is a Time versus Detector Signal (Intensity) plot

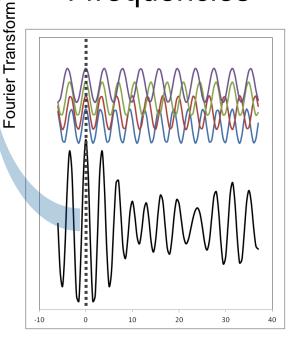
Detector Signal

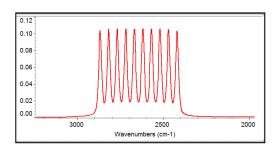


The FTIR Interferogram

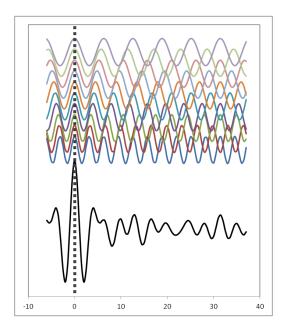


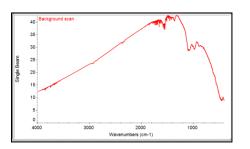
4 frequencies



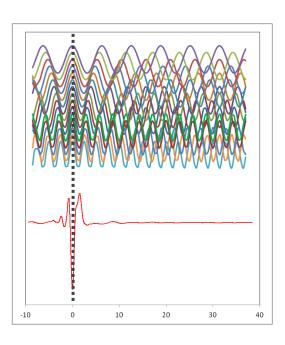


10 frequencies



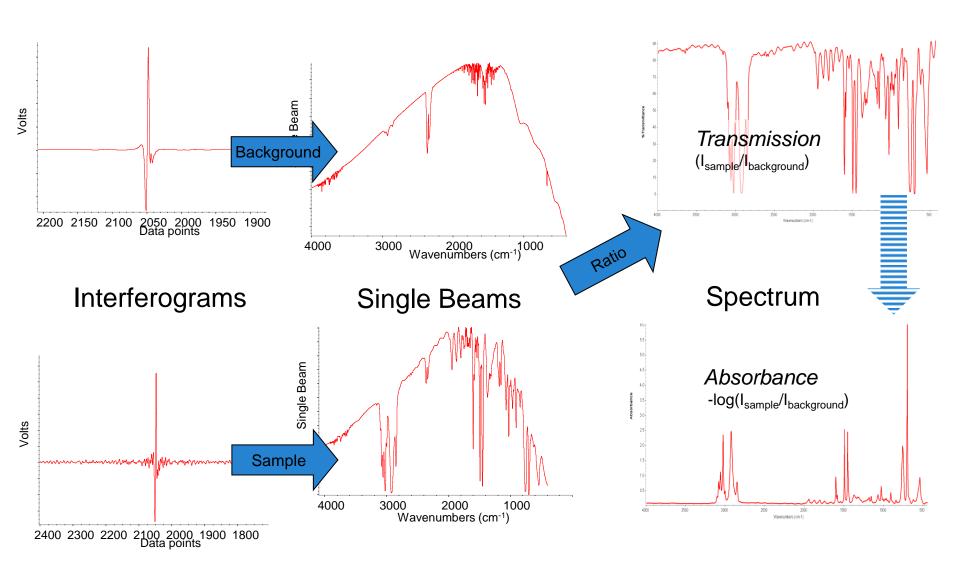


Broadband source

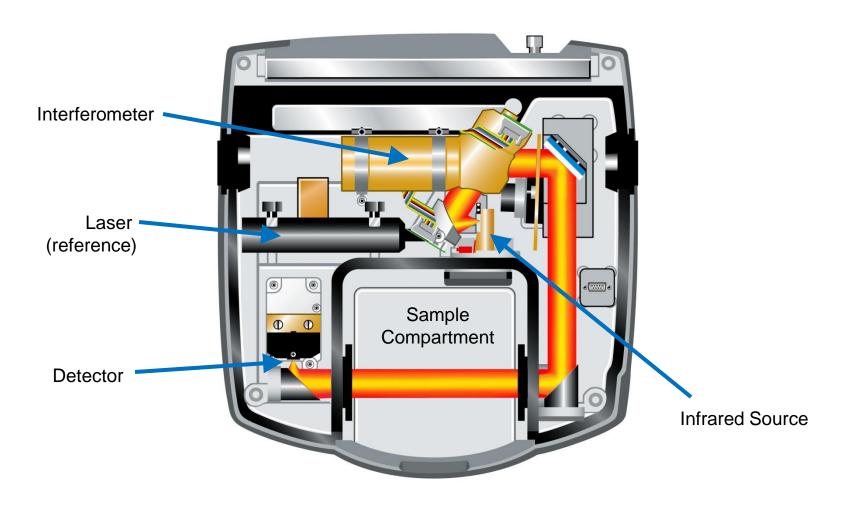




Raw Data to Final Spectrum



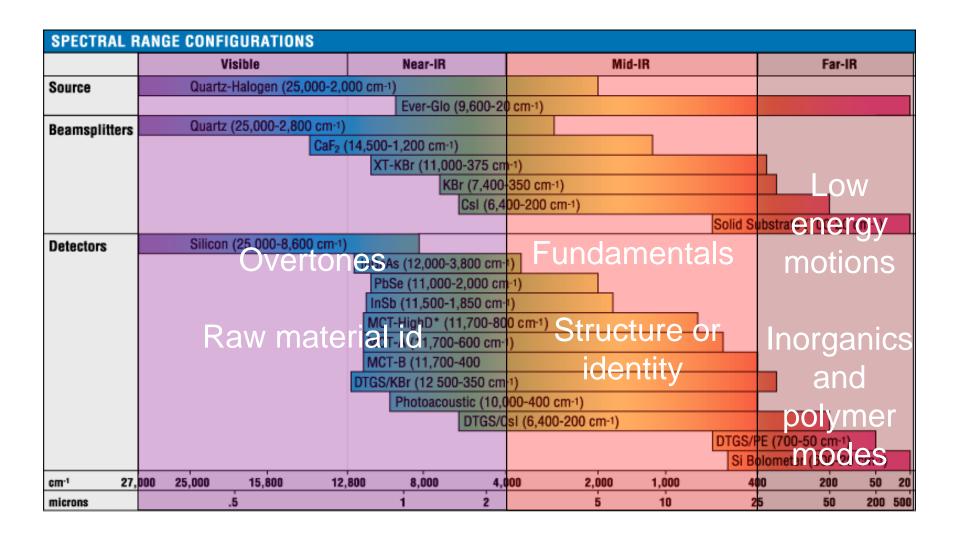
Inside the FT-IR Spectrometer



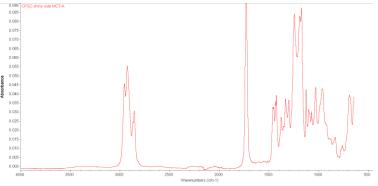
Thermo Scientific™ Nicolet™ iS™10 FT-IR Spectrometer Simplicity of form, dependability of function



Choose Components to Set Spectral Range



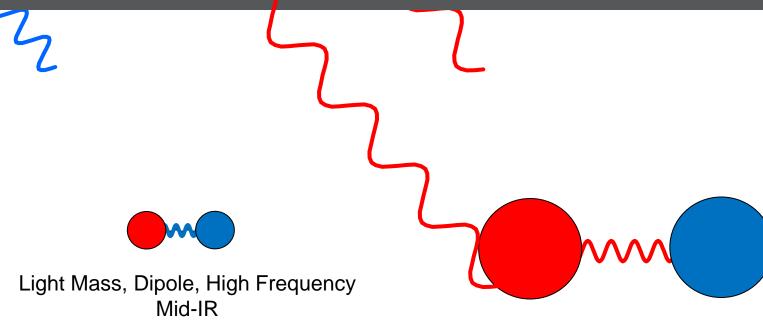
The Rest is Up to Physics!



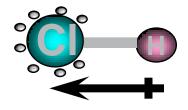


- The dipole moment must change
- Molecules absorb specific frequencies
 - Chemically similar materials absorb in the same range
 - Esters: around 1750 cm⁻¹
 - Hydrocarbons: around 3000 cm⁻¹
- Multiple uses
 - Identification
 - Quantitation
 - Kinetic studies

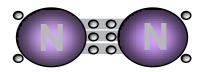
Basics: What Happens at the Sample?



Heavy Mass, Dipole, Low Frequency Far-IR



Only when there is a dipole change do you get IR peaks



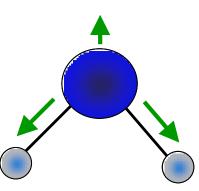
No Dipole, Nothing Happens

NIR: A powerful branch of the FT-IR Family

- NIR heavily used in industrial processes
 - Raw material identification (RMID)
 - "HOVal" for condensation analyses
 - Excellent for moisture detection
- Good Mid-IR generally means good NIR
 - NIR not covered today ask for more information!

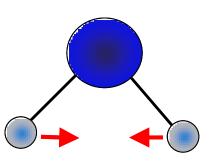


3600 cm⁻¹



5200 cm⁻¹

1600 cm⁻¹



Combination band in Water



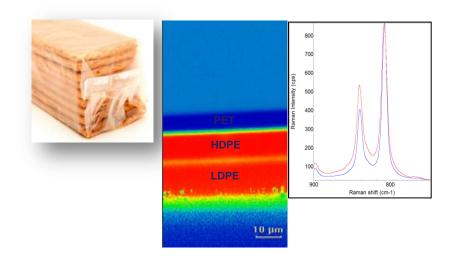


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Raman Spectroscopy

Basics: Raman as a Laboratory Tool

- Raman is used to solve problems
 - Non-destructively, non-invasively
 - Identify, map, image, depth profile
- Often coupled with a microscope
- Minimal or no Sample Preparation
- High Spatial Resolution
- What do you need?
 - Single action, repeated analyses?
 - Flexibility?
 - Imaging?



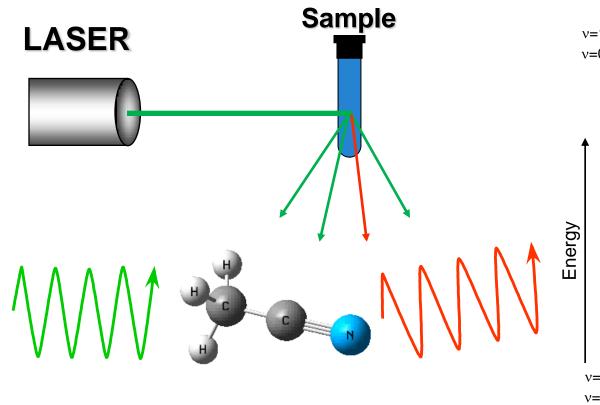


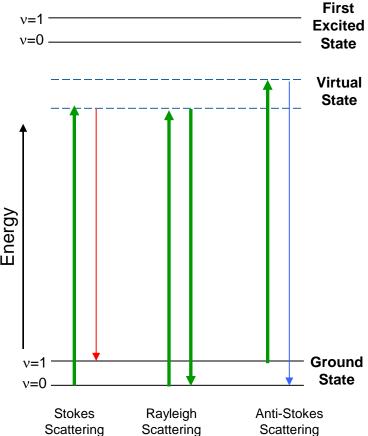


Basics: Select the Right Tool for your Application



Basics: Raman Spectroscopy





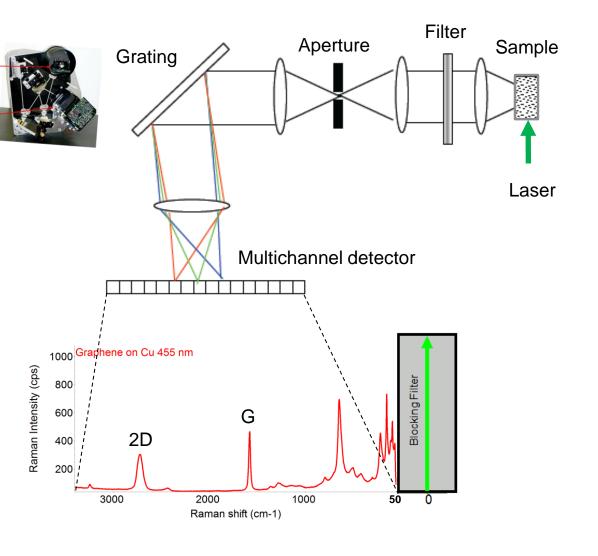
- Raman is a scattering technique
 - Sensitive to vibrational modes of covalent bonds in molecules
 - Most sensitive to highly symmetrical vibrations (i.e. C-C, Si-Si)
 - Sensitive to anything that changes bond energy (i.e. bond angle)



Basics: How Dispersive Raman Spectrometers Work

DXR2 System Basics:

- Lasers
 - 455 nm
 - 532 nm
 - 633 nm
 - 785 nm
- Rayleigh rejection filter
 - 50 cm⁻¹ cut-off
- Apertures
 - Slit
 - Pinhole (confocal mode)
- Gratings
 - Standard, 5 cm⁻¹
 - High Resolution, 2 cm⁻¹
- Detectors
 - CCD, EMCCD



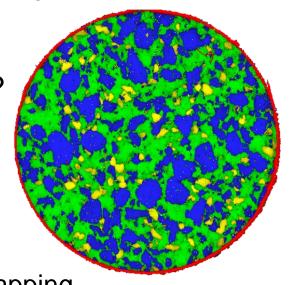


Basics: Why so Many Options on the DXR2?

- Multiple Lasers (455, 532, 633 or 785 nm)?
 - Trade-off: Fluorescence versus Efficiency
 - Fluorescence can be very strong, >> Raman
 - Efficiency strongly depends upon excitation wavelength

Efficiency $\propto 1/\lambda^4$

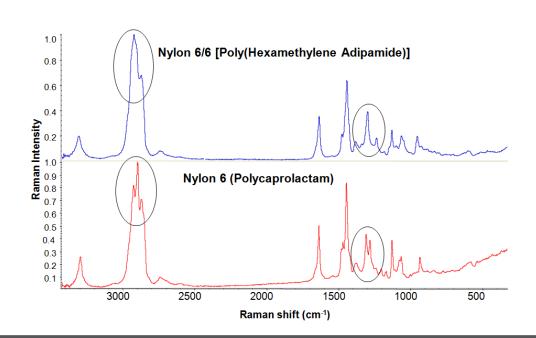
- High resolution versus standard resolution?
 - See closely spaced peaks, but...
 - Changes your available spectral range
- Detectors?
 - CCD: Standard detector, single point and mapping
 - EMCCD: Faster (not more sensitive), for fast imaging



The Rest is Up to Physics (again)!

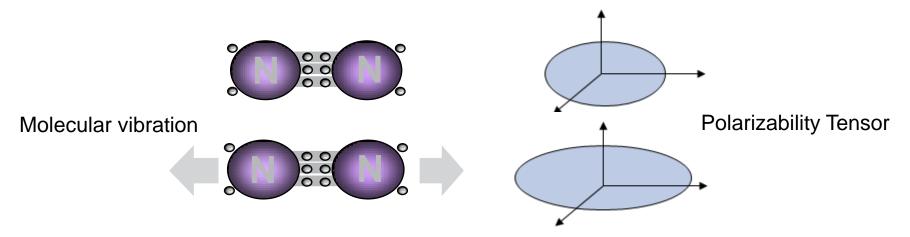
- There must be a polarizability change
 - Basically, a change in the electron cloud
- Peaks occur at specific frequencies
- Peaks may display a polarization dependence
 - Orientation and molecular symmetry
- Multiple uses
 - Identification
 - Quantity
 - Kinetics



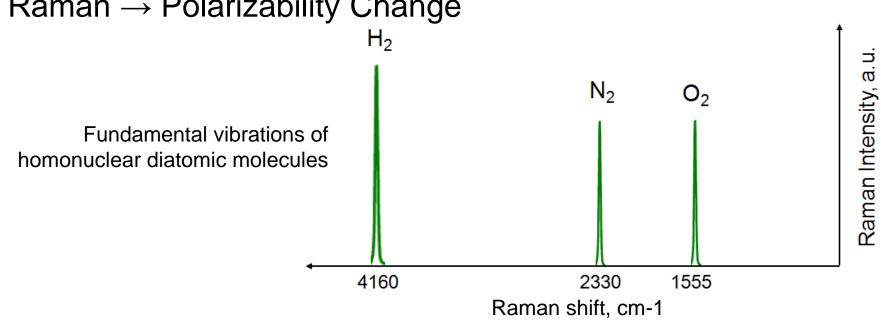




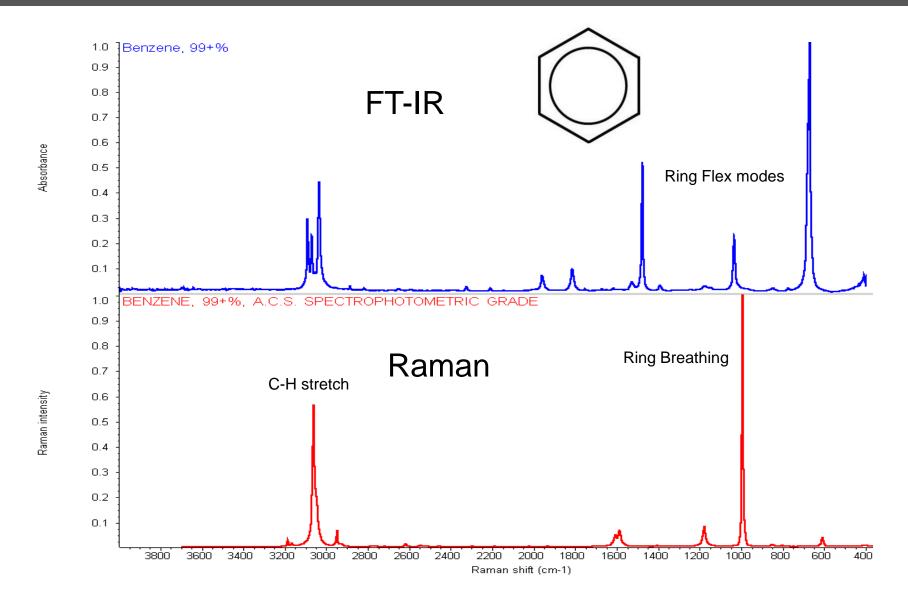
Basics: What Happens at the Sample?



- IR → Dipole Change
- Raman → Polarizability Change

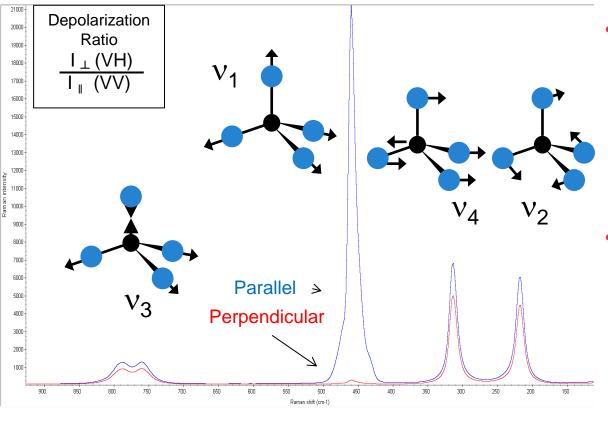


FT-IR and Raman Spectrum of Benzene





Additional Information: Raman Polarization



- Used to characterize:
 - Molecular symmetry
 - Molecular orientation
 - Crystallinity
 - Morphological traits
- DXR2 Raman polarization
 - Automated laser polarization
 - Automated analyzer orientation



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Extending Molecular Spectroscopy

Microspectroscopy and Multimodal Techniques

Basics: Microspectroscopy

 Combine FT-IR or Raman with microscope

 Probe content, homogeneity, contaminants

- FTIR or Raman
 - Today's last talk

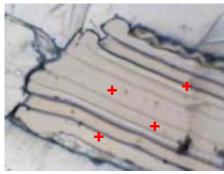
Point and shoot

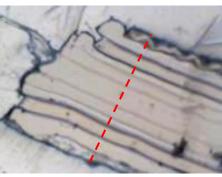


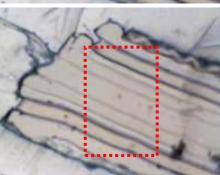
Cross-Sections



Chemical Images







Multimodal Techniques

- Relate Properties to Chemistry
 - Rheo-Raman or Rheo-IR
- Deformulate
 - TGA-IR
 - GC-IR
- Materials Analysis
 - XPS-Raman
 - AFM-Raman



Versatility: IR Microscopy, FT-Raman, Built-in ATR, automated Beamsplitter Exchanger, External experiment module



Versatility: The Thermo Scientific iXR Compact Raman Coupled with XPS



Molecular Spectroscopy: Research to Routine

- Discover.
 - Relate Viscosity to Chemistry
 - Biological imaging
- Solve.
 - Identify unknowns
 - Deformulate materials

Assure

- QA/QC on incoming, outgoing material
- In-line process control

